

- Measurement and source apportionment of PM_{2.5} were performed during the 2015 Spring Festival (SF).
- Vehicular emission and secondary aerosol formation were identified as the dominant sources before the SF.
- During the SF fireworks were the largest contributor to the $PM_{2.5}$, accounting for 30%.
- Secondary aerosols became predominant after the SF with a contribution of 44% to the PM_{2.5}.
- Fireworks may pose a serious health problem to local residents.

2	in urban Beijing during the 2015 Spring Festival			
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Characterization and source identification of fine particulate matter

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20 Abstract

21 The Spring Festival (SF) is the most important holiday in China for family reunion and 22 tourism. During the 2015 SF an intensive observation campaign of air quality was conducted to 23 study the impact of the anthropogenic activities and the dynamic characteristics of the sources. 24 During the study period, pollution episodes frequently occurred with 12 days exceeding the Chinese Ambient Air Quality Standards for 24-h average PM_{2.5} (75 µg/m³), even 8 days with 25 exceeding 150 μ g/m³. The daily maximum PM_{2.5} concentration reached 350 μ g/m³ while the 26 27 hourly minimum visibility was less than 0.8 km. Three pollution episodes were selected for detailed analysis including chemical characterization and diurnal variation of the PM2.5 and its 28 chemical composition, and sources were identified using the Positive Matrix Factorization model. 29 The first episode occurring before the SF was characterized by more formation of SO₄²⁻ and NO₃⁻ 30 and high crustal enrichment factors for Ag, As, Cd, Cu, Hg, Pb, Se and Zn and six categories of 31 pollution sources were identified, whereby vehicle emission contributed 35% to the PM2.5. The 32 second episode occurring during the SF was affected heavily by large-scale firework emissions, 33 which led to a significant increase in SO_4^{2-} , Cl⁻, OC, K and Ba; these emissions were the largest 34 contributor to the $PM_{2.5}$ accounting for 30%. During the third episode occurring after the SF, SO_4^{2-} , 35 NO_3^- and NH_4^+ and OC were the major constituents of the $PM_{2.5}$ and the secondary source was the 36 37 dominant source with a contribution of 44%. The results provide a detailed understanding on the variation in occurrence, chemical composition and sources of the PM2.5 as well as of the gaseous 38 39 pollutants affected by the change in anthropogenic activities in Beijing throughout the SF. They 40 highlight the need for limiting the firework emissions during China's most important traditional 41 festival.

- 42
- 43 Keywords: PM_{2.5}, firework emissions, source apportionment

44 1. Introduction

45 With the rapid development of economy, industrialization and urbanization during recent decades, air pollution has become a serious problem, which needs to be solved in China. Fine 46 47 particulate matter (particulate matter with aerodynamic diameter less than 2.5 µm, PM_{2.5}) is the primary pollutant in most cities and attracts worldwide attention because of adverse effects on 48 49 public health, visibility, air quality, and climate change (Chen et al., 2017; Cohen et al., 2017; Hu 50 et al., 2015a; Lin et al., 2016; Liu et al., 2014b). In recent years, haze, which is closely related to 51 elevated PM_{2.5} concentrations, occurs frequently throughout China, causing serious health impacts, 52 economic losses, and public complaints. So far, a series of the strictest measures for emission 53 control have been issued and implemented to resolve the air pollution problem by the Chinese 54 government from October 2013 (http://english.mep.gov.cn/News service/news release/index 26.shtml). The annual mean $PM_{2.5}$ concentrations in 2015 ranged from 22 to 107 μ g/m³ among the 55 74 key cities with an average value of 55 μ g/m³, exceeding the Chinese Ambient Air Quality 56 Standards (CAAQS) Grade II annual mean of 35 µg/m³ (Beijing environmental statement, 2015). 57 The Beijing-Tianjin-Hebei region (BTH) is one of the most polluted areas, with an annual average 58 $PM_{2.5}$ concentration of 77 µg/m³ in 2015, approximately 2.2 times the CAAQS II threshold and 59 60 5.1 times the secondary air quality standard of the United States (annual mean: 15 μ g/m³) 61 (https://www.epa.gov/pm-pollution/table-historical-particulate-matter-pm-national-ambient-air-qu ality-standards-naaqs). Thus, this region still faces severe challenges to mitigate PM2.5 levels to 62 63 meet both national and international regulations.

64 Beijing is the core city in the BTH region with the number of local residents being 65 approximately 22 million and 273 million tourists visiting Beijing in 2015. As the capital city of China, the severe air pollution is of great concern because of its adverse health impacts on 66 millions of local residents and foreigners working and travelling here (Guo et al., 2017). 67 68 Significant emission control efforts have been made especially targeting key sources of concern to 69 the reduce air pollution levels in Beijing. Although large improvement in air quality was made during important events, including the 2008 Olympic Games, the 2014 Asia-Pacific Economic 70 71 Cooperation conference and the 2015 China Victory Parade Day (Jia et al., 2011; Wang et al., 2017; Wang et al., 2016; Xie et al., 2017), the PM_{2.5} concentration did not decline substantially in 72

autumn and winter and the annual mean concentration was still approximately 80 μ g/m³ in 2015 (Beijing environmental statement, 2015). In particular, Beijing suffered 42 days of severe air pollution with an average value of 239 μ g/m³ in 2015. This indicated that current control measures might not be effective and that emission control efforts need to focus more accurately on the predominant sources during the pollution episodes. To further improve the air quality, it is necessary to characterize the PM_{2.5} pollution episodes and to identify the sources of the PM_{2.5}.

79 Numerous studies have been initiated to investigate the evolution of the chemical 80 composition of PM_{2.5} and to identify the sources in Beijing (Hu et al., 2015b; Yu et al., 2013; Zhang et al., 2013). However, most source apportionment studies were based on continuous filter 81 82 sampling (12 h or 24 h) and offline laboratory analysis (Liu et al., 2017a; Wang et al., 2016; Yao et 83 al., 2016; Zong et al., 2016). The results reflected the average source contributions over different 84 observation periods but could not shed light on the detailed variation of the air pollutants and the 85 dynamic characteristics of the sources during severe pollution episodes, which last for short periods of less than one week. Thus, it is of great significance to analyze PM_{2.5} and its chemical 86 87 compositions in detail during severe episodes using near real-time data from state-of-the-art 88 devices such as the Aerosol Mass Spectrometers (Elser et al., 2016), rapid collection of 89 particles-ion chromatography systems (RCFP-IC) (Wen et al., 2006), semi-online OC and EC 90 analyzers (Ji et al., 2016) and online multi-metal monitors (Gao et al., 2016), which used in 91 combination allow one to determine the sources of the $PM_{2.5}$ (Gao et al., 2016; Jeong et al., 2016).

92 The Spring Festival (SF) is the most important holiday in China for family reunion and 93 tourism. Extremely heavy traffic always occurs during this period because plenty of migrants, who 94 work in megacities of China, go to their hometown before the SF and afterwards return back to 95 work (Huang et al., 2012); besides, Beijing attracts approximately 10 million of tourists during the 96 SF (http://travel.people.com.cn/n/2015/0226/c41570-26599014.html). Some of the industrial 97 activities are temporarily halted so that the energy consumption decreases accordingly. However, 98 intensive fireworks are held for celebration of the festival and they give rise to degradation in air 99 quality (Huang et al., 2012; Ji et al., 2017; Kong et al., 2015). Some studies have focused on the 100 effect of the SF holiday on the change in criteria air pollutants (SO₂, NO_x, CO, O₃, PM_{2.5} and PM₁₀) 101 (Dong et al., 2014; Jin et al., 2007) or the chemical composition of PM_{2.5} (Feng et al., 2016; 102 Huang et al., 2012; Kong et al., 2015; Tian et al., 2014; Wang et al., 2015). However, most of the

103 results solely focused on the variation in gaseous pollutants, PM or a special category of chemical 104 species of $PM_{2.5}$ such as water-soluble inorganic ions, carbonaceous aerosols or trace elements and 105 they made use of filter sampling and offline analysis. Thus, the variation in chemical composition 106 and sources, which resulted from the dramatic change in anthropogenic activities including the 107 emissions from the fireworks in a short-term period could not clearly be identified due to the low 108 time resolution of the data or inadequate chemical information (Zou and Yao, 2014).

109 To fill this research gap, $PM_{2.5}$ and its chemical species including water-soluble inorganic 110 ions, carbonaceous species and trace elements as well as major gaseous pollutants were 111 concurrently measured on an hourly basis in Beijing from February 8 to March 2, 2015. The 112 variation in $PM_{2.5}$ and its chemical species were determined, the sources were identified and the 113 impact of the fireworks was evaluated. To our knowledge, this is the first time that near real-time 114 instruments were jointly used to measure PM_{2.5} and its chemical species in urban Beijing during 115 the SF. The results provide valuable information for more effective PM_{2.5} control in Beijing or 116 other megacities in China during holidays.

117 **2. Methodology**

118 2.1. Sampling site and observation period

119 The sampling site (39°58′28″N, 116°22′16″E) is located inside the campus of the State Key 120 Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry, Institute of 121 Atmospheric Physics (IAP), Chinese Academy of Sciences, in urban Beijing and is surrounded by 122 busy urban roads (Fig. S1). The number of automotive vehicles in Beijing amounts to approximately 5.62 million and the average vehicular speeds within the urban traffic network were 123 124 28.1 and 25.1 km/h in 2015 during the morning and evening rush hours, respectively (All the data 125 in this study on traffic, such as the traffic congestion index were obtained from 126 http://www.bjjtw.gov.cn/bmfw /dljtyxqk/201503/t20150303_94290.html). Considering that a large 127 number of industrial enterprises have been moved out of urban Beijing or phased out, there were no immediate industrial sources near the sampling site. However, small industrial sources and 128 129 their possible operating schedules are unknown. The experimental campaign was conducted from 130 February 8 to March 2, 2015, whereas the SF lasted for 7 days (from February 19 to 25). Air samples were collected at approximately 8 m above the ground. As shown in Table S1, the PM_{2.5} 131 132 concentrations at this site were significantly correlated with those at the 10 urban national sites,

including the Olympic center (Aoti), the Agriculture exhibition center (Nongzg), Dongsi (Dongs),
the Temple of heaven (Tiant), Huayuan (Huay), Wanliu (Wanl), Gucheng (Guc), Guanyuan
(Guany), Wanshouxigong (Wansxg) and Dingling (Dingl), which suggests that the sampling site is
representative for urban Beijing. The locations of all the above sites were introduced in Cheng et
al. (2015).

138 2.2. Instrumentation

To monitor $PM_{2.5}$ use was made of a synchronized hybrid ambient real-time particulate analyzer (Model 5030, Thermo-Fisher Scientific, Massachusetts, USA), which has been approved as a Federal Equivalent Method analyzer by the US EPA. The minimum detectable concentration limit was less than 0.2 μ g/m³ for 24 h. Operation, calibration and maintenance were strictly performed according to the instruction manual (https://tools.thermofisher.com/content/sfs/ manuals/EPM-manual-Model%205030%20SHARP.pdf).

145 A RCFP-IC system was used to monitor SO_4^{2-} , NO_3^{-} , CI^- , NH_4^+ , K^+ , Na^+ , Ca^{2+} and Mg^{2+} 146 hourly in PM_{2.5}. A semi-continuous organic carbon (OC) and elemental carbon (EC) field analyzer 147 (Model 4, Sunset Laboratory Inc., Oregon, USA) with a modified NIOSH-like protocol was 148 employed to obtain hourly OC and EC concentrations in PM_{2.5}. Detailed descriptions of the 149 RCFP-IC system and the semi-continuous OC and EC analyzer were given in Wen et al. (2006) 150 and Ji et al. (2016), respectively.

151 Hourly trace elements concentrations in PM2.5 were obtained with an automated multi-metals 152 monitor (Xact 625, Cooper Environmental Services, Oregon, USA). The quantification of the 153 metals including K, Ca, Fe, Ag, Ba, As, Cd, Cu, Cr, Mn, Hg, Ni, V, Se, Zn, Ga, Pb, Co, Sn, Sb, Pd, Au and Tl in PM2.5 was based on the US EPA method IO 3.3 using nondestructive 154 155 energy-dispersive X-ray fluorescence. Operation, calibration and maintenance of the instrument 156 were strictly performed according to its instruction manual (https://archive.epa.gov/nrmrl 157 /archive-etv/web/pdf/p100cxcl.pdf). The stability of the automated multi-metals monitor was 158 checked during every sampling and analysis cycle using an internal Pd rod.

Atmospheric temperature (T), relative humidity (RH), wind direction (WD) and wind speed
(WS) were obtained using an automatic weather station (Model MILOS 520, VAISALA Inc.,
Finland) and atmospheric visibility was recorded via a visibility sensor (Model 6000, Belfort
Instrument, Maryland, USA).

163 **2.3. Source apportionment**

Positive Matrix Factorization (PMF), which is an effective mathematical receptor model maintained by US EPA, is used worldwide for source apportionment. The principle and methodology of PMF have been reviewed and described in detail elsewhere (Han et al., 2017; Zíková et al., 2016; Zhang et al., 2013). The goal of the PMF model is to minimize the objective function *Q*:

169
$$Q = \sum_{i=1}^{n} \sum_{j=1}^{m} \left[\frac{x_{ij} - \sum_{k=1}^{p} g_{ik} f_{kj}}{u_{ij}} \right]^2$$

Where x_{ij} and u_{ij} are the concentration and uncertainty of the *jth* species in the *ith* sample, 170 171 f_{ki} is the fraction of the *jth* species in *kth* source, g_{ik} indicates the contribution of the *kth* source to the *ith* sample, p is the number of sources, and m and n are the total number of species and 172 173 samples, respectively. In this study, EPA's PMF version 5.0 (Norris et al., 2014) was used for 174 source apportionment of the PM2.5. Ca and K were measured by the automated multi-metals monitor and Ca²⁺ and K⁺ by the RCFP-IC system, but the results from the former were selected 175 176 for the source apportionment in this study. Thus, the PMF input dataset included PM_{2.5}, Cl⁻, Na⁺, Mg²⁺, NH₄⁺, SO₄²⁻, NO₃⁻, OC, EC, K, Ca, Fe, Ag, As, Cd, Ba, Cr, Cu, Mn, Ni, Hg, V, Se, Pb, and 177 178 Zn. PM_{2.5} was set as total variable. More information on the PMF as used in this study is given in 179 the Supplementary material. The Q-values, source profiles and scaled residuals distributions were 180 examined to obtain the most reasonable factor solution. The uncertainties of the PMF results were 181 assessed by the displacement (DISP) and bootstrap (BS) error estimation methods with 100 BS 182 runs and a minimum Pearson correlation coefficient of 0.6. In addition, the F_{peak} function was used 183 to control the rotational ambiguity.

184 3. Results and discussion

185 **3.1.** Variations in air pollutants during the study period

Figs. 1(a) and 1(b) depict the hourly $PM_{2.5}$, SO_2 , CO, NO_2 and O_3 concentrations and the hourly values of T, RH, WS, WD and visibility at the sampling site. As shown in Fig. 1(a), there were five typical cyclic accumulation processes of $PM_{2.5}$ recorded during the study period (Ji et al., 2012). High $PM_{2.5}$ concentrations were observed when southerly wind or calm wind prevailed while low concentrations were recorded during cold fronts accompanied by strong wind arriving in Beijing. The hourly concentrations of $PM_{2.5}$ varied from 3 to 430 µg/m³ with an average of

 $115\pm107 \ \mu\text{g/m}^3$. The daily averaged PM_{2.5} concentrations ranged from 17 to 350 $\mu\text{g/m}^3$. There 192 were 12 days for which the CAAQS 24-h Grade II standard for $PM_{2.5}$ (75 µg/m³) was exceeded, 193 even 8 days for which 150 μ g/m³ was exceeded. The highest hourly and daily PM_{2.5} 194 195 concentrations were recorded on February 15 before the SF; they were attributable to a combined effect of unfavorable meteorological conditions, heavy traffic caused by plenty of migrants going 196 197 to their hometown (From February 4 to 15 the peak traffic congestion index was approximately 198 6.0, indicating that almost all major roads were congested) and residential heating. Compared to 199 other studies conducted during SF periods, the daily average $PM_{2.5}$ concentration of 115±90 µg/m³ 200 in this study was slightly higher than those in Shanghai (Huang et al., 2012), Nanjing (Kong et al., 201 2015) and Xinxiang (Feng et al., 2016), but lower than those in Tianjin (Tian et al., 2014), Zhengzhou (Liu et al., 2014a) and Jinan (Yang et al., 2014) (Table S2). The hourly PM_{2.5} 202 203 concentrations at the IAP showed a similar variation as those observed in Tianjin, Shijiazhuang 204 and Xingtai, which are major cities in the BTH region (Fig. S2), indicating that pollution episodes 205 of a similar pattern occurred in different cities within the region during this period.

SO₂, CO and NO₂ showed a similar temporal variation as PM_{2.5}. The maximum hourly CO 206 concentration, 6.6 mg/m³, was observed on February 15 morning while that of NO₂ occurred on 207 208 February 14 night, possibly because of continuing unfavorable meteorological conditions (low WS 209 and high RH) and increased emissions from the high traffic intensity. The hourly maximum SO_2 concentration of 121 µg/m³ was found at 1:00 on New Year's Eve (February 18); it was associated 210 211 with firework emissions (Huang et al., 2012). Table S3 shows that SO_2 , CO and NO_2 were significantly correlated with PM2.5. SO2 exhibited a high positive correlation with PM2.5, 212 suggesting that coal combustion and the firework emissions contributed a lot to the PM2.5 (Ji et al., 213 214 2017). In contrast, O₃ was negatively correlated with the PM_{2.5}. That is because high PM_{2.5} 215 concentrations are usually accompanied with accumulation of primary pollutants (such as NO), 216 which could consume O_3 (Han et al., 2011).

The whole campaign was divided into three periods: before the SF (from February 8 to 17), during New Year's Eve and the SF (from February 18 to 25) and after the SF (from February 26 to March 2). The average $PM_{2.5}$ mass concentrations and associated standard deviations before, during and after the SF were 138 ± 123 , 121 ± 94 and $61\pm12 \ \mu g/m^3$, respectively. The highest $PM_{2.5}$ concentration was recorded before the SF, followed by during and after the SF, consistent with the decreasing strengths of anthropogenic emissions throughout the sampling period (Huang et al., 2012). That the lowest $PM_{2.5}$ level occurred after the SF could be due to favorable meteorological conditions and low emissions from anthropogenic sources since many migrant workers did not return to work immediately after the SF and the construction and industrial activities were not yet at their normal operational levels. A total of five pollution episodes was identified for the study period. They are listed in Table S4 of the Supplementary material; three of them were selected for detailed analysis in the following section.

229 **3.2.** Chemical characteristics of the three selected episodes

230 To better understand the evolution and sources of the air pollution throughout the festival period, three episodes were selected for further analysis as shown in Fig. 1(a) (shaded and gridded 231 232 areas). The first episode (E1) was recorded from February 8 to10 before the SF. The second 233 episode (E2) was observed from February 17 to 19 during the SF (February 18 was the Chinese 234 New Year's Eve and February 19 was the first day of the SF). The third episode (E3) was recorded from February 27 to March 1 after the SF, when the holiday was ending and migrant workers and 235 students gradually returned, which reflects the transition period from slightly to fully operational 236 237 industrial activities. Note that the pollution episode occurring during February 13-15 was not 238 selected as the multi-metals monitor did not work properly then.

239 **3.2.1. E1 before the SF**

Fig. 2 shows the hourly concentrations of the PM_{2.5} constituents during E1. From 15:00 on 240 February 8, NO₃⁻, NH₄⁺, SO₄²⁻, Cl⁻, OC and EC started to rise and they peaked at approximately 241 1:00 on February 10. The highest concentrations of SO_4^{2-} , NO_3^{-} , NH_4^{+} , Cl⁻, OC and EC reached 27, 242 39, 23, 11.9, 57 and 9.0 μ g/m³, respectively. During the rapid increase stage, the increment rates 243 of NO₃, NH₄⁺, SO₄²⁻, Cl⁻, OC and EC were 0.7, 0.5, 0.4, 0.3, 1.4 and 0.2 μ g/m³/h, respectively. 244 245 OC had the highest rate and a significant correlation between OC and EC with a slope of 6.3 was 246 recorded, which reflects a combined contribution from vehicular exhaust, coal combustion, and 247 biomass burning (Cao et al., 2005). Usually the SF travel rush starts 1-2 weeks before the SF and 248 intensive traffic activities result in high OC and NO₃⁻ contributions to the PM_{2.5}, as observed in previous research (Huang et al., 2012; Kong et al., 2015). The significant correlation between OC 249

and CO (R^2 =0.81) further corroborates the impact of vehicle emissions.

The atmospheric transformation of SO_2 to SO_4^{2-} and of NO_2 to NO_3^- could be calculated by the sulfur oxidation ratio (*SOR*) and nitrogen oxidation ratio (*NOR*) based on equations (1) - (3):

(2)

253
$$SOR = n[nss-SO_4^{2-}]/(n[SO_2] + n[nss-SO_4^{2-}])$$
 (1)

 $[nss-SO_4^{2-}] = [SO_4^{2-}] - 0.2517 \times [Na^+]$

255
$$NOR = n[NO_3^-]/(n[NO_2] + n[NO_3^-])$$
 (3)

where $nss-SO_4^{2-}$ denotes non-sea-salt sulfate and the unit for *n* is mole/m³; 0.2517 is the ratio of SO₄²⁻ to Na⁺ in sea water (Tao et al., 2009). During the rapid increase stage, *SOR* and *NOR* were 0.13 and 0.17, suggesting rapid gas-particle transformation (Zhang et al., 2016). The NO₃^{-/}SO₄²⁻ mass ratio can be used to discuss the contributions of stationary and mobile sources (Feng et al., 2016). The average NO₃^{-/}SO₄²⁻ ratio of 1.6±0.7 suggests that the vehicle contribution was significantly larger than that from coal combustion during E1; this is consistent with the result obtained by PMF shown in Fig. 5.

K, Ca, Fe and Zn were the major elemental constituents with average concentrations of 263 1,340±360, 158±58, 360±110 and 240±100 ng/m³, respectively (Table 1). Before 15:00 on 264 265 February 8, all elements showed opposite variation patterns compared with the pattern for secondary inorganic aerosol (see Fig. 2). This could mainly be due to the significant contribution 266 267 of elements from windblown dust (Shen et al., 2016), as is supported by high wind speeds shown 268 in Fig. 1(b). Whereas most crustal matter and urban road dust contains a K/Fe ratio of less than 1.0, 269 much higher ratios were noted during the evening and early morning hours (approximately 5:1) 270 and they were still elevated during the day (approximately 2:1) (Chan et al., 1997). This might 271 suggest a contribution from biomass burning at that time. During the accumulation process, Pb, Zn, 272 As and Se showed an increased contribution with average concentrations of 98 ± 6 , 240 ± 100 , 29 ± 10 and 5.3 ± 1.4 ng/m³, respectively; these elements are closely related with combustion 273 274 (Pokorná et al., 2016). It suggests that coal combustion used for heating was still important during 275 this episode.

The degree of enrichment relative to soil crustal matter for trace elements can be quantitatively assessed by the crustal Enrichment Factor (*EF*) based on equation (4):

278
$$EF = (X/X_{Ref})_{aerosol}/(X/X_{Ref})_{soil}$$

279 where X is element of interest and X_{Ref} is the reference element in soil (Huang et al., 2012; Liu et

(4)

280 al., 2017a). In this study, Fe in background soil in China was chosen as the reference element (Wei et al., 1991). The average EF values for elements in the three episodes are given in Table S5. 281 282 Elements with EF values smaller than 10, which are usually considered to originate primarily from natural sources (Xu et al., 2013) included K, Fe, Ca, Cr, Mn and Ba during E1. 283 Anthropogenic elements including Cu, As, Zn, Pb, Se, Ag, Hg and Cd had EF values higher than 284 285 100, ranging from 168 (Cu) to 10100 (Cd), suggesting the importance of anthropogenic sources. Note that the EF values of the elements declined during the rapid increase stage except for Ag and 286 287 Cd, suggesting that these elements were not associated with the main cause of the PM_{2.5} increase. Additionally, it was found that the EF values of Ni and V were higher than 10 during 4:00-7:00 on 288 289 February 8 and 8:00-17:00 on February 10; this could be due to a substantial contribution from 290 heavy oil combustion for these two elements (Dall'Osto et al., 2013).

291 **3.2.2. E2 during the SF**

292 Fig. 3 shows the temporal variation of the PM2.5 constituents during E2 and Table 1 presents 293 their average concentrations and associated standard deviations. Due to increased wind speed from the northwest or northeast varying from 2.2 to 5.2 m/s and because of a decline in anthropogenic 294 295 emissions (i.e., from industries and vehicles), the concentrations PM2.5 and its chemical species 296 declined before 18:00 on February 18. The average visibility was 46 ± 6 km. With the shift of WD 297 from north to south and the decrease of WS, a PM_{2.5} peak was recorded from 22:00 on February 17 to 2:00 on February 18 and as a result SO₄²⁻, NO₃⁻, NH₄⁺, Cl⁻, OC and EC peaked. Afterwards, 298 the hourly PM_{2.5} concentrations began to decrease until 18:00 on February 18 (New Year's Eve). 299 Then the PM_{2.5} concentrations increased quickly from 33 to 400 μ g/m³ and peaked at 1:00 on 300 February 19 when huge amounts of firework emissions occurred. The time period with PM_{2.5} 301 concentrations of more than 75 μ g/m³ lasted for 70 h after the fireworks. 302

The highest concentrations of OC, EC, Cl⁻, NO₃⁻, SO₄²⁻, K⁺, NH₄⁺, Na⁺ and Mg²⁺ were 20, 2.1, 57, 12.6, 92, 96, 9.1, 2.2 and 5.9 μ g/m³ at 1:00 on February 19. K⁺, SO₄²⁻ and Cl⁻ increased at fast rates of 13.3, 11.5 and 7.7 μ g/m³/h, respectively. These increases result from the fact that firework materials are made up of sulfur powder, carbon powder and KClO₄ (Ji et al., 2017). KClO₄ acts as the major oxidizer and the corresponding chemical reaction (R1) results in high Cl⁻ and SO₄²⁻ concentrations during the firework. In addition, potassium sulfate is also used as an additive in the firework (Perry, 1999).

$$310 KClO4 + S + C \rightarrow SO2 + K2SO4 (SO2) + KCl + CO2 (CO) (R1)$$

(5)

311
$$\sum$$
Anions = [SO₄²⁻]/48 + [NO₃⁻]/62 + [Cl⁻]/35.5 + [F⁻]/19

312
$$\sum \text{Cations} = [\text{NH}_4^+]/18 + [\text{Mg}^{2+}]/12 + [\text{Ca}^{2+}]/20 + [\text{Na}^+]/23 + [\text{K}^+]/39$$
 (6)

The Σ Cations/ Σ Anions molar ratio was 0.97 for the period from 19:00 on February 18 to 313 1:00 on February 19, which indicated that there was a slight deficiency in cations to neutralize the 314 anions. A NO₃⁻ peak did not occur, in contrast to what was the case for K^+ , SO₄²⁻ and Cl⁻. This is 315 because NO₃⁻ is not a major constituent of firework materials and the meteorological conditions 316 were not favorable for secondary formation of NO_3 . Note that the PM, which originates from 317 318 fireworks can remain in the atmosphere for a long time and may be a substrate for heterogeneous reactions of SO₂ and NO₂, thus degrading air quality when favorable meteorological conditions 319 prevail (Kong et al., 2015). As reported by Pathak et al. (2009), NO_3^- is preferably formed in an 320 321 ammonia-deficient atmosphere via (R S4) in the Supplementary material, for which basically 322 heterogeneous hydrolysis of N2O5 occurs on the moist surface of aerosols. The PM emitted by the fireworks during the SF provides an appropriate interface to form NO_3^- via heterogeneous 323 324 reactions when the RH increases.

The correlations between OC and CO and between OC and EC were significant during E2. 325 326 Furthermore, high OC/EC ratios (on average 9.2 ± 5.3) were observed, which possibly result from 327 the adhesive organic matter in the firework materials and in addition from vehicular exhaust and 328 coal combustion. As shown in Fig. 3, K, Ba, Ca, Fe, Pb and Cu increased at rates of 870, 14.2, 10.1, 9.4, 4.7 and 4.7 ng/m³/h, respectively, during the intensive firework emission period (from 329 330 19:00 on February 18 to 1:00 on February 19). The EF values of K, Ba, Cu and Pb reached 29, 22, 331 129 and 210, respectively, at 1:00 on February 19. The Ba, Cu and Pb concentrations increased by 332 factors of 3.4, 5.6 and 1.8, respectively, in a short time, which could pose health risks to the local 333 residents, in particular to children (Hamad et al., 2016). The average Ba concentration (99 ng/m³) 334 during the intensive emission period was higher than that observed in Nanjing during the Spring Festival (70 ng/m³) (Kong et al., 2015). The results indicate that fireworks lead to substantial 335 336 emissions of its related elements.

337 **3.2.3. E3 after the SF**

As shown in Fig. 4, SO₄²⁻, Cl⁻, NO₃⁻, NH₄⁺, OC and EC varied similarly to PM_{2.5} with 338 average concentrations of 15.4 \pm 15.8, 2.6 \pm 2.2, 12.0 \pm 12.8, 12.2 \pm 13.9, 12.6 \pm 9.7 and 1.6 \pm 1.4 μ g/m³, 339 respectively. Compared to E2, the NO_3^- and NH_4^+ concentrations were much higher during E3. 340 341 After the SF, migrant workers gradually returned to Beijing for work and traffic flows and 342 industrial activities increased progressively. The emission intensities of various sources changed and the ratios of NO_3^{-} , SO_4^{-2-} and NH_4^{++} to total $PM_{2,5}$ varied correspondingly. The ratios of 343 $NO_3^{-}/PM_{2.5}$, $SO_4^{-2}/PM_{2.5}$ and $NH_4^{+}/PM_{2.5}$ were 0.18±0.08, 0.29±0.21 and 0.17±0.08, respectively. 344 During E3, the ratio of Σ Cations/ Σ Anions was 1.4 (with R^2 =0.98 between both sums), indicating 345 346 that the PM_{2.5} was alkaline, which was different from that during the intensive firework emission period. A significant correlation was found between OC and EC was with a slope of 6.4 and a R^2 347 of 0.87. The high OC/EC ratios were related to the combined contributions from vehicular exhaust 348 349 and coal combustion (Cao et al., 2005).

The average concentrations of K, Fe, Zn, and Ca were 410 ± 170 , 87 ± 40 , 36 ± 19 and 34 ± 18 ng/m³, respectively, and aggregately these four elements accounted for 88.8% of the mass of all elements in this episode. As shown in Table S5, the *EF* values of Cd, Ag, Hg, Se, Pb, Zn, Cu, Ni, As and V were all higher than 10, indicating a quite important contribution for these elements from anthropogenic emissions. In contrast to what was the case in E2, the average *EF* values of elements related to industries like Cr, Cu, Zn, Hg and Pb increased during E3, which seems to reflect the progressive re-opening of the industrial activities.

The diurnal variation in $PM_{2.5}$ and its chemical constituents during E1, E2 and E3 was also examined; it is shown in the Supplementary material (Figs. S3, S4 and S5). Due to the influence of different source strengths, the diurnal variation was in accordance with the morning and evening rush hours and nighttime emissions as well as the additional emission from the fireworks during the SF.

362 **3.3.** Source apportionment of PM_{2.5} for the three selected episodes

The PMF model was run 100 times with a random seed and the lowest Q was considered as base run solution. After examination of the F_{peak} values, the base run results ($F_{peak} = 0$) were retained for the three selected episodes. Mg²⁺ was excluded from the input of the model for E1 and E3 because of lower signal-to-noise ratios and Ag and Cd were excluded for all three selected episodes due to the lower coefficients between measured and predicted values ($R^2 < 0.01$). Cr and Na⁺ were also excluded for E2 and E3 because of poor match between the modeled and measured values, respectively. The PMF solutions with from four to eight factors were examined. The detailed solution analysis is presented in the Supplementary material. Seven, six and six factors were found to be reasonable for E1, E2 and E3, respectively.

372 Fig. 5 presents the seven identified source profiles and their average contributions to the 373 PM_{2.5} for E1. Factor 1 with high loadings of Ca, Cr, Mn, Fe and Zn was identified as dust because 374 of the presence of the crustal elements Ca, Mn, and Fe. Ba, Cu, As, Se, Hg and Pb were also found 375 in this factor. Road dust, windblown dust and construction dust could not be distinguished from each other because of the insufficient tracers for dust. Dust was an important source of PM2.5 with 376 an average contribution of 5.8%. Factor 2 was termed a source associated with vehicle exhaust 377 because it is characterized by high loadings of NO₃⁻, NH₄⁺, OC and EC (Liu et al., 2017b; Singh et 378 379 al., 2016; Zong et al., 2016). It is well known that vehicular emissions provide a substantial contribution to the OC and EC levels and that NH_4^+ is emitted by vehicles that are equipped with 380 381 three-way catalytic converters (Chang et al., 2015). This source contributed with 35% to the PM₂₅ 382 during E1, which is in accordance with the SF travel rush. Factor 3 was identified as coal combustion, as it is characterized by high Cl⁻, OC and EC loadings. The high Cl⁻ loading 383 384 associated with coal combustion in winter is a distinctive feature in Beijing and inland China (Liu 385 et al., 2017b; Zhang et al., 2013). The contribution of coal combustion to the PM_{2.5} was 21%. Factor 4 can be determined as secondary sources because of the high loadings of NO_3^- , NH_4^+ , and 386 SO_4^{2-} . This factor provided a contribution of 29% to the PM_{2.5}. Factor 5 has high loadings for V 387 388 and Ni, which are related with oil combustion (Wang et al., 2016); it contributed with 2.1% to the 389 $PM_{2.5}$. The sixth factor stood for industrial sources; it is characterized by a higher loading of Cu, 390 Zn and As as well as a moderate loading of Hg, Pb and Cr (Wang et al., 2016; Yao et al., 2016). Since there are no industries in Beijing and because of the temporary shutdown of manufacturing 391 392 in the surrounding areas of Beijing around the time of the SF, the industrial source only 393 contributed with 3.9% during E1. Factor 7, characterized by a high loading of K and accounting for 4.2% of the PM_{2.5}, was biomass burning and waste incineration (Betha et al., 2014). As, Se, Hg 394 395 and Pb also had a moderate loading in factor 7; the latter elements are also related with biomass

burning and waste incineration (Dall'Osto et al., 2013).

397 For E2, the six-factor solution provided the most physically reasonable source profiles. 398 Factor 1 was considered as vehicular exhaust, as it is characterized by high loadings of NO₃, NH₄⁺, OC and EC (see Fig. 6); it contributed with 26% to the PM_{2.5}. Factor 2 had high loadings of 399 NO_3^{-} , NH_4^{+} and SO_4^{-2-} and was therefore identified as secondary sources but it is also partly related 400 401 to aging processes of firework emissions because it shows some loading of Ba, Cu and K (Kong et 402 al., 2015). This source contributed with 17.8% to the PM2.5. Factor 3 was dust, characterized by 403 high Ca, Mn, Fe, Se, Ba and Pb, and it provided a contribution of 5%. Note that dust is partly 404 associated with the firework emissions because they cause soil dust resuspension. Factor 4 was 405 coal combustion, characterized by high Zn, As, Se, Hg and Cl⁻; it has also a moderate loading of 406 OC and EC. As well known, Hg, As, Se and Zn originate from coal combustion (Feng et al., 2016; Huang et al., 2012; Wang et al., 2010). This factor contributed with 11.5% to the PM_{2.5}. Compared 407 408 to E1, a reduction of approximately 10% was found for coal combustion during E2; this is 409 attributed to the decline in industrial activities during the SF. Factor 5 is termed firework emissions, as it is characterized by high Cl⁻, SO₄²⁻, Mg²⁺, OC, EC, K and Ba (Huang et al., 2012; 410 411 Kong et al., 2015; Yang et al., 2014); it contributed with 30% to the $PM_{2..5}$. This factor stood out 412 as the most significant contributor. Factor 6 might be associated with oil combustion for residential heating, as it is characterized by high V and Ni. This factor accounted with 9.3% to the 413 414 PM_{2.5}.

415 The source profiles of the six factors and their contributions for E3 are presented in Fig. 7. Factor 1 could be categorized as vehicle exhaust since it has elevated levels of OC and EC. NO₃ 416 417 and NH_4^+ also have some moderate contribution to factor 1. Vehicle exhaust contributed with 32% to the PM2.5 during E3. Factor 2 was coal combustion as it is characterized by high loadings of OC, 418 419 Cl⁻ and EC (Zhang et al., 2013; Liu et al., 2017b); it accounted for 13.4% of the PM_{2.5}. Factor 3 420 was oil combustion characterized by high Ni and V loadings and it accounted for 4.3% of the 421 PM_{2.5}. Factor 4 is considered as an industrial source, since it is characterized by high Cr, Mn, Fe, 422 Cu, and Zn. It contributed with 5.2% to the PM_{2.5} whereas no industrial source was present for E2; 423 its presence in E3 is attributed to the re-opening of most industrial factories. Factor 5 was secondary sources, characterized by NO3⁻, SO4²⁻ and NH4⁺; it accounted for 44% of the PM2.5. 424 425 Factor 6 could be considered as biomass burning and waste incineration, because of the high K

426 loading. Moderate loadings of As, Se, Hg and Pb are also seen in this factor. This factor 427 contributed with 1.6% to the PM_{2.5}.

428 As shown in Fig. 8, substantial differences in source contributions between the three episodes 429 were found. The vehicle contribution was highest (35%) during E1; firework emissions were most 430 important during E2, when they provided a contribution of 30%, while the secondary source was 431 the largest contributor (44%) during E3. The hourly contribution of the different sources during 432 the episodes is shown in Figs. S6, S7 and S8; the patterns are consistent with the emission sources 433 that were identified above. Finally, an analysis based on chemical mass closure was also used for 434 apportioning the particulate matter to its components; this work is given in the Supplementary 435 material (Figs. S9 and S10).

436 Conclusions

An intensive observation campaign was organized to study the air quality in urban Beijing during the 2015 SF. $PM_{2.5}$ and its chemical species, including water-soluble inorganic ions, carbonaceous species and trace elements were measured with 1-h time resolution. Based on the chemical characterization and source apportionment analysis, the following conclusions can be drawn:

442 (1) During the Spring Festival period the changes in anthropogenic activities played an important 443 role in the variation in air pollutants and severe pollution episodes occurred frequently. Before the 444 SF, the higher concentrations of CO, NO₂ and $PM_{2.5}$ were related to a combined effect of 445 unfavorable meteorological conditions, heavy traffic, and residential heating. During the SF the 446 firework emissions gave rise to a steep increase in $PM_{2.5}$ and SO_2 for a short time. After the SF, 447 increased human activities and regional transport contributed a lot to the $PM_{2.5}$ loading.

448 (2) During the three selected episodes secondary inorganic ions and carbonaceous species 449 accounted for most of the $PM_{2.5}$. Secondary conversion acts a pivotal process in the formation of 450 air pollution. The firework emissions released lots of gaseous pollutants and particles containing 451 harmful chemicals like heavy metals in a short time, posing health risks to tourists and local 452 residents. Thus, more attention should be paid to limit fireworks during the SF.

(3) The results of the source apportionment indicated that vehicle exhaust, fireworks andsecondary sources, respectively, were important contributors during the three selected episodes,

455 which is consistent with the change in emission sources.

This study enhances our understanding of the impact of the change in anthropogenic activities on the characteristics, formation mechanisms, and sources of the air pollution. Based on near real-time data of the chemical composition of the $PM_{2.5}$, the evolution of the air pollution episodes, their dynamical characteristics and the sources could be precisely identified. Future work will focus on developing methods for monitoring more tracers of the different emission sources with high time resolution to improve the accuracy of the source apportionment.

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Table 1 Average concentrations and associated standard deviations of $PM_{2.5}$ (µg/m³) as well as of water-soluble inorganic ions, carbonaceous species (µg/m³) and trace elements (ng/m³) in PM_{2.5} during the three selected episodes of the study period.

Species	E 1	E2	E3
PM _{2.5}	107±70	102±99	72±76
Water-soluble	inorganic ions		
Cl	5.2±3.2	11.5±15.2	2.6±2.2
NO ₃ ⁻	18±12	9.9±9.0	12±13
SO_4^{2-}	11±6.1	23±25	15±16
Na^+	1.8±0.3	1.4 ± 0.4	1.6±0.3
\mathbf{K}^+	1.0±0.6	18±25	1.2±1.2
$\mathrm{NH_4}^+$	11±6.7	7.3±8.1	12±14
Mg^{2+}	0.1±0.0	$1.1{\pm}1.7$	0.1±0.0
Carbonaceous	species		
OC	26±16	14±12	13±9.7
EC	3.8±2.4	$1.9{\pm}1.7$	1.6±1.4
Trace elements	5		
Κ	1350±370	1350±1100	410±172
Ca	158±49	156±119	34±18
V	5.7±10	16±7.7	7.1±7.0
Cr	3.0±2.1	0.5 ± 0.8	0.4±0.6
Mn	29±12	10±4.7	4.1±2.6
Fe	360±110	250±130	87±40
Ni	3.0±2.6	5.8±2.1	3.2±2.7
Cu	44±23	11±6.2	5.1±2.0
Zn	240±100	45±18	36±19
As	29±10	8.1±2.6	2.3±3.3
Se	5.3±1.4	2.0±0.8	0.6±0.4
Ag	6.8±3.7	7.8±3.0	6.7±2.5
Cd	11±3.1	11±3.2	9.9±2.5
Ba	32±11	59±65	9.6±5.6
Hg	3.6±0.7	$2.7{\pm}1.0$	1.5±0.3
Pb	98±26	39±15	18±14

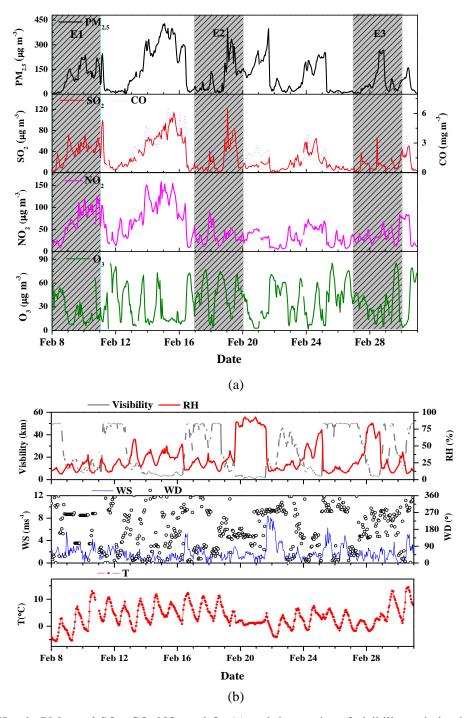


Fig. 1 Hourly $PM_{2.5}$ and SO_2 , CO, NO_2 and O_3 (a) and time series of visibility, relative humidity, wind speed, wind direction and temperature (b) from February 8 to March 2, 2015 at the observation site.

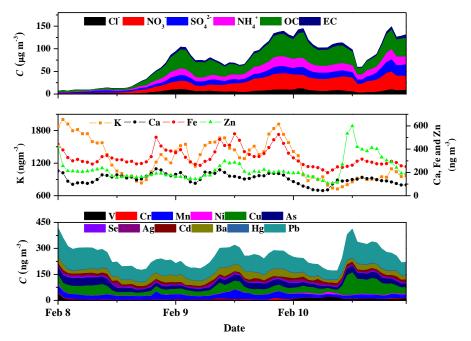


Fig. 2 Hourly concentrations of chemical species in $\ensuremath{\text{PM}_{2.5}}$ during E1.

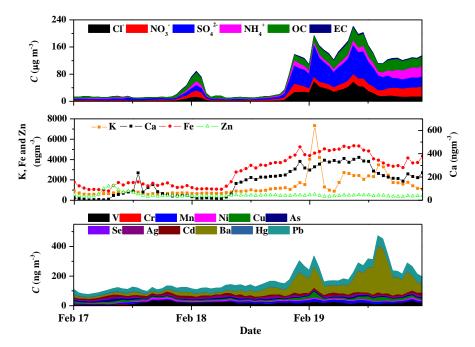


Fig. 3 Hourly concentrations of chemical species in $\ensuremath{\text{PM}_{2.5}}$ during E2.

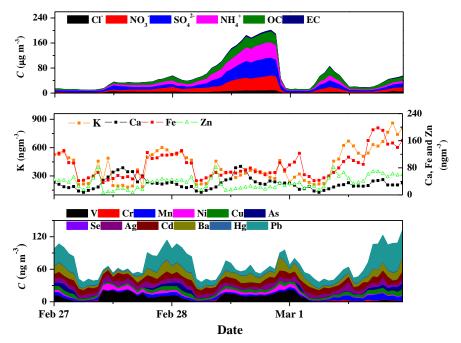


Fig. 4 Hourly concentrations of chemical species in $\ensuremath{\text{PM}_{2.5}}$ during E3.

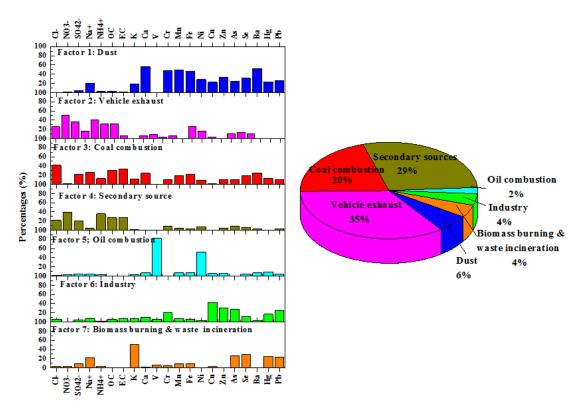


Fig. 5 Profile and contribution of each source during E1.

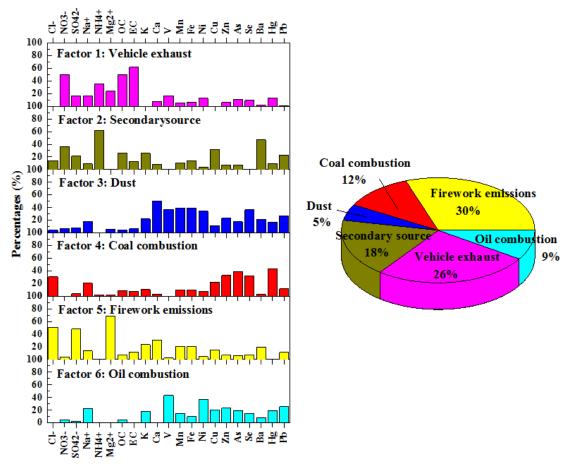


Fig. 6 Profiles and contribution of each source during E2.

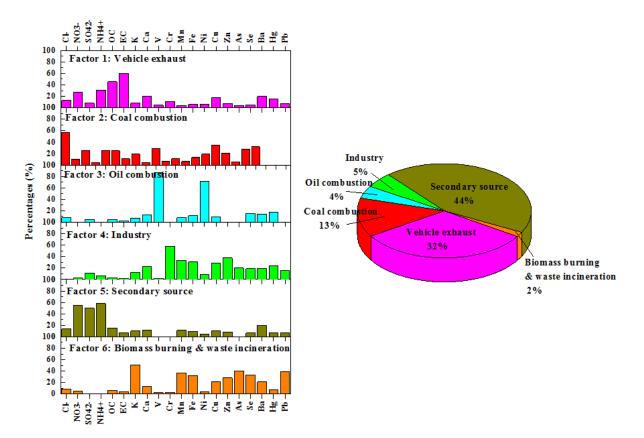


Fig. 7 Profile and contribution of each source during E3.

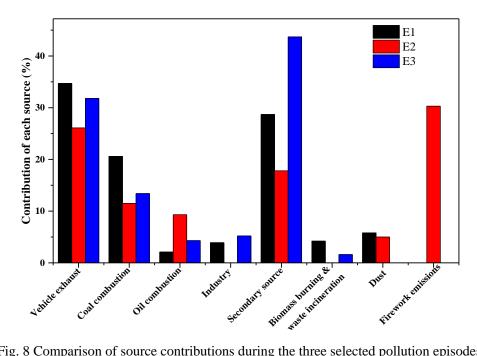


Fig. 8 Comparison of source contributions during the three selected pollution episodes.

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